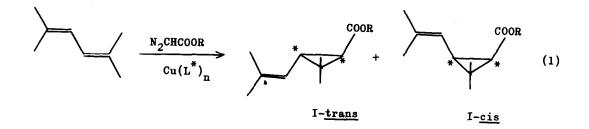
Tetrahedron Letters No. 21, pp 1707-1710, 1975. Pergamon Press. Printed in Great Britain.

ASYMMETRIC SYNTHESIS OF CHRYSANTHEMIC ACID. AN APPLICATION OF COPPER CARBENOID REACTION

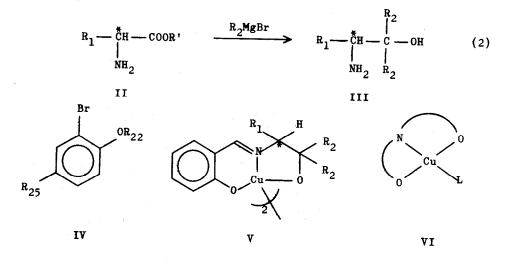
T. Aratani, Y. Yoneyoshi, and T. Nagase Central Research Laboratory, Sumitomo Chemical Co., Ltd. Takatsuki, Osaka, Japan

(Received in Japan 31 March 1975; received in UK for publication 11 April 1975)

Asymmetric induction was observed in the decomposition of diazoalkanes in the presence of certain chiral copper complexes.¹ We have applied this chiral carbenoid reaction to the synthesis of chrysanthemic acid² (eq. 1). Systematic studies have been carried out to achieve an optical yield of 60-70%.³



A catalyst is prepared as follows. (<u>L</u>)-Alanine ethyl ester (II, $R_1 = Me$, R' = Et) was allowed to react (70°, 5 hr) with a Grignard reagent (5 molar excess) derived from 2-bromo-4-<u>t</u>-butylphenyl octyl ether (IV, $R_{22} = octyl$, $R_{25} = \underline{t}$ -Bu) in tetrahydrofuran to give (<u>S</u>)-2-amino-1,1-di(5-<u>t</u>-butyl-2-octyloxyphenyl)-1-propanol (III, $R_1 = Me$, $R_2 = 5-\underline{t}$ -butyl-2-octyloxyphenyl), a viscous oil, $[\alpha]_D$ -25.5° (<u>c</u> 1.80, chloroform) in 73% yield after column chromatography (silica gel, benzene-ether).⁴ Treatment of an ethanolic solution of salicylaldimine of the amino alcohol III with an equimolar amount of cupric acetate followed by alkaline workup afforded a binuclear copper (II) complex (S)-V ($R_1 = Me$, $R_2 = 5-\underline{t}-butyl-2-octyloxyphenyl$), a green mass, (αI_{546} -550° (<u>c</u> 0.06, benzene) in 86% yield.⁵ Upon addition of pyridine the complex V was converted to a mononuclear complex VI (the same 0,N,0 ligand as in V; L = pyridine).



Ethyl diazoacetate was decomposed (40°, 5 hr) in 2,5-dimethyl-2,4-hexadiene (4 molar excess) in the presence of the catalyst (\underline{S})-V (1 mol %) to give a mixture of <u>cis</u>- and <u>trans</u>-ethyl chrysanthemate (I, R = Et), b.p. 60°/0.5 mm, a_{D} -13.52° (neat, 1 dm) in 54% yield. Hydrolysis of the ester gave the acid I (R = H), b.p. 80°/0.5 mm, $[al_{D}$ -38.76° (\underline{c} 5.10, chloroform), whose optical isomers were analyzed by gas chromatography as the corresponding (-)-2-octyl ester⁶: (-)-<u>trans</u>,43.3%; (+)-<u>trans</u>, 8.1%; (-)-<u>cis</u>, 39.4% and (+)-<u>cis</u>, 9.2%. The enantiomeric excess is calculated to be 68% for <u>trans</u> isomer and 62% for <u>cis</u> one. It should be noted that the pyridine adduct VI is as effective as the binuclear complex V in the asymmetric synthesis.

Other catalysts V were prepared in the combination of optically active a-amino ester II ($R_1 = Me, \underline{i}-Pr, \underline{i}-Bu, CH_2Ph$, and Ph; R' = Et) and 2,5disubstituted phenyl bromide IV (R_{22} , $R_{25} = Me, \underline{i}-Pr, \underline{n}-Bu, \underline{t}-Bu$ and so on). Results of the asymmetric synthesis are summarized in Table I, which reveals the following facts. First, the catalyst V of (S)-configuration leads to the predominant formation of levorotatory chrysanthemic acid (both of (1S,3S)-<u>trans</u> and (1S,3R)-<u>cis</u> isomer) over dextrotatory one. On the other hand, dextrotatory chrysanthemic acid is the preferred product in the case of the catalyst (R)-V. Second, the optical activity of the product increases with the bulkiness of the substituent R₂ (R₂₂ and R₂₅) of V.⁷

A copper-carbene complex such as VI (L = CHCOOR or N_2 CHCOOR) might be responsible for the asymmetric induction.⁸ Experiments directed on this line are now in progress.

	Cataly	lyst V		Ester I $(R = Et)$			Acid I $(R = H)$
Confign	R ₁	R ₂₂	R ₂₅	Yield,% ^b	<u>cis/trans</u> ^c	a _D (neat) ^d	[a] ^e
(<u>s</u>)	Me	Me	н	64	38/62	-3.74	-11.02
(<u>s</u>)	Ме	<u>n</u> -Bu	Me	60	41/59	-7.22	-20.46
(<u>s</u>)	Me	<u>n</u> -Bu	<u>t</u> -Bu	54	42/58	-11.58	-35.84
(<u>R</u>)	CH ₂ Ph	Me	Н	58	38/62	+8.01	+23.10
(<u>R</u>)	CH ₂ Ph	<u>i</u> -Pr	н	53	42/58	+8.80	+25.48
(<u>R</u>)	CH ₂ Ph	<u>n</u> -Bu	<u>t</u> -Bu	52	43/57	+10.26	+29.84
(<u>s</u>)	<u>i</u> -Pr	<u>n</u> -Bu	<u>t</u> −Bu	59	40/60	-9.67	-24.60
(<u>s</u>)	<u>i</u> -Bu	<u>n</u> -Bu	<u>t</u> -Bu	46	42/58	-5.65	-16.90
(<u>R</u>)	Ph	Me	Н	51	33/67	+1.18	+3.54

Table I. Asymmetric Synthesis of Chrysanthemic Acid^a

^a For the reaction conditions, see the text.

b Based on ethyl diazoacetate.

^C Determined on gas chromatography (Golay column).

^d Measured with a cell of 1 dm length at room temperature.

^e The concentration <u>c</u> (g/100 ml chloroform) was <u>ca</u>. 5. Optically pure acid shows the rotation of $[a]_{D}$ +26.8° for (+)-<u>trans</u> isomer and that of $[a]_{D}$ +87.9° for (+)-<u>cis</u> one. <u>Acknowledgement</u>: The authors are grateful to Miss M. Okuno and Dr. S. Nakamura for their assistance and help.

REFERENCES

- H. Nozaki, H. Takaya, S. Moriuti, and R. Noyori, <u>Tetrahedron</u>, <u>24</u>, 3655 (1968);
 R. Noyori, H. Takaya, Y. Nakanisi, and H. Nozaki, <u>Can. J. Chem.</u>, <u>47</u>, 1242 (1969);
 T. Aratani, Y. Nakanisi, and H. Nozaki, <u>Tetrahedron</u>, <u>26</u>, 1765 (1970).
- M. Matsui and I. Yamamoto, "Naturally Occurring Insecticides" (edited by M. Jacobson and D. G. Crosby), Marcel Dekker, New York (1971).
- 3. Patents pending.
- For the reaction between optically active α-amino esters and phenylmagnesium bromide, see A. McKenzie, R. Roger, and G. O. Wills, <u>J. Chem. Soc</u>., 779 (1926); B. M. Benjamin, H. J. Schaeffer, and C. J. Collins, <u>J. Am. Chem. Soc</u>. <u>79</u>, 6160 (1957).
- A kind of Schiff base can behave as a tridentate ligand towards cupric ion:
 R. P. Houghton and D. J. Pointer, <u>J. Chem. Soc</u>., 4214 (1965); M. Kishita,
 Y. Muto, and M. Kubo, <u>Australian</u> J. <u>Chem.</u>, <u>10</u>, 386 (1957).
- 6. A. Murano, Agr. Biol. Chem. (Tokyo), 36, 2203 (1972).
- Similar observation was reported in the catalytic asymmetric hydrogenation:
 W. S. Knowles, M. J. Sabacky, and B. D. Vineyard, <u>Chem. Commun.</u>, 10 (1972).
- The copper(I) species might be involved in the system. Cf., R. G. Salomon and J. K. Kochi, <u>J. Am. Chem. Soc.</u>, <u>95</u>, 3300 (1973).
- <u>Addendum</u>. Very recently the effectiveness of bis(camphorquinonedioximato)cobalt(II) was demonstrated in the asymmetric synthesis of ethyl 2-phenylcyclopropanecarboxylate: Y. Tatsuno, A. Konishi, A. Nakamura, and S. Otsuka, <u>J. Chem. Soc., Chem. Commun</u>., 588 (1974).